

REACTIONS OF 1,3-SUBSTITUTED BENZOTHIENO[2,3-*c*]PYRYLIUM SALTS WITH PRIMARY AMINES

S. V. Tolkunov, M. A. Kryuchkov, V. S. Tolkunov, and V. I. Dulenko

*We have studied the reactions of benzothieno[2,3-*c*]pyrylium salts with primary amines. We have shown that recyclization occurs in two directions, and whether benzothieno[2,3-*c*]pyridinium salts or 1-aminodibenzothiophenes are formed is determined by the nature of the primary amine.*

Keywords: 1-aminodibenzothiophenes, benzothieno[2,3-*c*]pyridinium perchlorates, benzothieno[2,3-*c*]pyrylium perchlorates, recyclization.

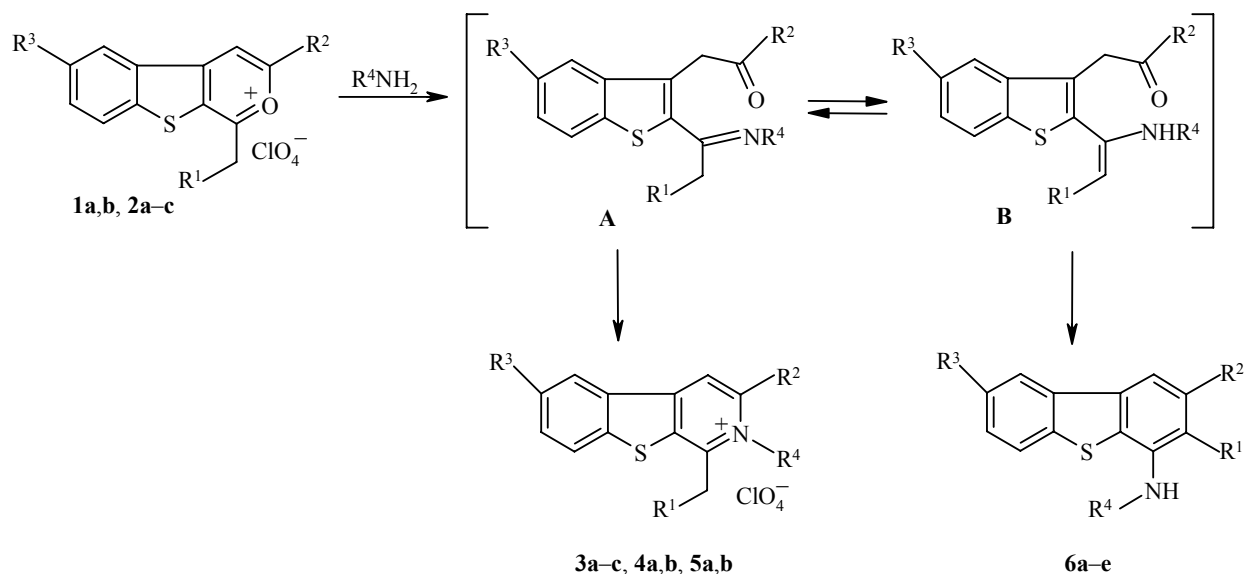
Recyclization of alkyl-substituted pyrylium salts by primary amines can occur both with participation of the α -alkyl substituents (leading to alkyylanilines) or without involvement of the substituents (forming N-pyridinium salts). The reaction of monocyclic pyrylium salts and 1,3-dialkyl-substituted benzo[*c*]pyrylium salts with primary amines leads to pyridinium and isoquinolinium salts [1, 2]. On the other hand, 1-alkyl-3-aryl- and 1-alkyl-3,4-diaryl-substituted benzo[*c*]pyrylium salts reacted with primary amines yield only substituted naphthylamines [3]. Reactions between primary amines and pyrylium salts annelated with heterocycles have not been studied.

We have studied the reaction of 1,3-disubstituted benzothieno[2,3-*c*]pyrylium salts **1**, **2** with primary amines. It has been hypothesized that recyclization of pyrylium salts by primary amines occurs according to an ANRORC scheme, and includes addition of the nucleophile at the position 1 of the pyrylium ring [4]. Further opening of the pyran ring leads to intermediate formation of ketimines **A**, which exist in equilibrium with the tautomeric enamine form **B**. Formation of the end products (benzothieno[2,3-*c*]pyridinium salts **3-5** and 1-aminodibenzothiophenes **6**) obviously is determined by the ratio of forms **A** and **B**, and depends on the nature of amine.

So reaction of pyrylium salts **1a,b** with anilines leads to pyridinium salts **3b,c** and **4a,b**, while recyclization of pyrylium salts **2a,b** by methylamine and monoethanolamine leads exclusively to the 1- R^4 -aminobenzothiophenes **6b-d**. Reactions of the salts **1a** and **2c,b** with benzylamine and furfurylamine lead to a mixture of pyridinium salts **3a**, **5a,b** and 1- R^4 -aminodibenzothiophenes **6a,d,e**.

The structure of 1- R^4 -aminobenzothiophenes **6a-e**, established from the ^1H NMR spectra, confirms that addition of primary amines to pyrylium salts **1**, **2** occurs at the position 1. The presence of an arylamino group at the position 3 of the pyrylium ring should signify that the product of ring opening after addition of amine at $C_{(1)}$ occurs to be not able to undergo further recyclizations. In fact, in studying the reaction of 1,6-dimethyl-3-(4'-tolylamino)benzothieno[2,3-*c*]pyrylium perchlorate (**7**) with benzylamine, we isolated for the first time the

L. M. Litvinenko Institute of Physical Organic Chemistry and Coal Chemistry, National Academy of Ukraine, Donetsk 83114; e-mail: tolkunov@uvika.dn.ua. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, No. 8, pp. 1246-1250, August, 2004. Original article submitted October 11, 2001.



1 a, b $R^1 = H, R^2 = Me, a R^3 = H, b R^3 = Cl$; **2 a, c** $R^1 = H, b R^1 = Me, a-c R^2 = Ph, a, b R^3 = H, c R^3 = Me$; **3 a-c** $R^1 = R^3 = H, R^2 = Me, a R^4 = Bn, b R^4 = p\text{-MeC}_6\text{H}_4, c R^4 = p\text{-ClC}_6\text{H}_4$; **4 a,b** $R^1 = H, R^2 = Me, R^3 = Cl, a R^4 = Ph, b R^4 = p\text{-ClC}_6\text{H}_4$; **5 a** $R^1 = H, b R^1 = Me, a, b R^2 = Ph, a R^3 = Me, b R^3 = H, a, b R^4 = \text{furfuryl}$; **6 a-c, e** $R^1 = H, d R^1 = Me, a R^2 = Me, b-e R^2 = Ph, a-d R^3 = H, e R^3 = Me; a R^4 = Bn, b R^4 = Me, c R^4 = \text{CH}_2\text{CH}_2\text{OH}, d, e R^4 = \text{furfuryl}$

TABLE 1. Characteristics of Synthesized Compounds

Compound	Empirical formula	Found, %					mp, °C	Yield, %
		Calculated, %						
		C	H	Cl	N	S		
3a	$C_{20}H_{18}ClNO_4S$	59.61	4.33	8.65	3.31	7.80	270	14
		59.48	4.46	8.80	3.47	7.93		
3b	$C_{20}H_{18}ClNO_4S$	59.36	4.31	8.94	3.33	7.85	243-244	78
		59.48	4.46	8.80	3.47	7.93		
3c	$C_{19}H_{15}Cl_2NO_4S$	53.61	3.41	16.88	3.45	7.61	267-268	74
		53.77	3.54	16.75	3.30	7.55		
4a	$C_{19}H_{15}Cl_2NO_4S$	53.95	3.41	16.86	3.47	7.60	283	58
		53.77	3.54	16.75	3.30	7.55		
4b	$C_{19}H_{14}Cl_3NO_4S$	49.91	3.17	23.35	3.12	7.12	273-274	50
		49.73	3.05	23.23	3.05	6.98		
5a	$C_{24}H_{20}ClNO_5S$	61.52	4.39	7.34	2.73	6.69	248	12
		61.34	4.26	7.56	2.98	6.82		
5b	$C_{24}H_{20}ClNO_5S$	61.52	4.18	7.73	2.81	6.77	173	18
		61.34	4.26	7.56	2.98	6.82		
6a*	$C_{20}H_{18}ClNS$	70.57	5.44	10.21	4.23	9.35	171-172	23
		70.69	5.30	10.46	4.12	9.43		
6b	$C_{19}H_{15}NS$	79.02	5.31	—	4.71	11.16	94-95	86
		78.89	5.19	—	4.85	11.07		
6c	$C_{20}H_{17}NOS$	75.17	5.47	—	4.28	10.21	148-149	90
		75.24	5.33	—	4.39	10.03		
6d*	$C_{24}H_{20}ClNOS$	71.15	4.87	8.61	3.60	7.67	175-177	57
		71.02	4.93	8.75	3.45	7.89		
6e	$C_{24}H_{19}NOS$	78.22	5.37	—	3.65	8.82	116	78
		78.05	5.15	—	3.79	8.67		
8	$C_{27}H_{26}N_2OS$	75.87	6.25	—	6.44	7.60	150-152	91
		76.02	6.14	—	6.57	7.52		

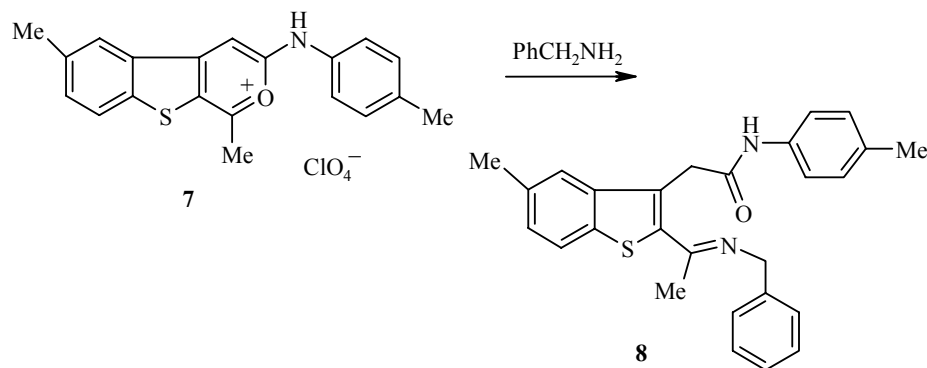
* Hydrochlorides.

TABLE 2 ¹H NMR Spectral Characteristics of Benzothieno[2,3-*c*]pyridinium perchlorates **3-5** and 1-*R*⁴-aminodibenzothiophenes **6a-e**

Com- pound	Chemical shifts, δ , ppm (<i>J</i> , Hz)			
	Aliphatic protons			Aromatic protons
	1-CH ₃ (1-C ₂ H ₅) [4'-CH ₃]	3-CH ₃ (6-CH ₃) [2-CH ₃]	NR ⁴ , NHR ⁴	
3a	3.03	2.84	6.03 (CH ₂)	7.10 (2H, m, 2'-, 6'-H); 7.40 (3H, m, 3'-, 4'-, 5'-H); 7.78 (1H, t, 7-H); 7.90 (1H, t, 6-H); 8.35 (1H, d, <i>J</i> ₅₆ = 8.4, 5-H); 8.65 (1H, d, <i>J</i> ₇₈ = 8.4, 8-H); 8.94 (1H, s, 4-H)
3b	3.70 [2.28]	2.55	—	7.44 (2H, d, <i>J</i> ₃₂ = 8.2, 3'-, 5'-H); 7.69 (2H, d, <i>J</i> ₂₃ = 8.2, 2'-, 6'-H); 7.76 (1H, t, 7-H); 7.88 (1H, t, 6-H); 8.40 (1H, d, <i>J</i> ₈₇ = 8.0, 8-H); 8.75 (1H, d, <i>J</i> ₅₆ = 8.0, 5-H); 9.05 (1H, s, 4-H)
3c	2.69	2.50	—	7.76-8.00 (2H, m, 6-, 7-H); 7.79 (2H, d, <i>J</i> ₃₂ = 8.2, 3'-, 5'-H); 7.90 (2H, d, <i>J</i> ₂₃ = 8.2, 2'-, 6'-H); 8.40 (1H, d, <i>J</i> ₈₇ = 8.0, 8-H); 8.75 (1H, d, <i>J</i> ₅₆ = 8.0, 5-H); 9.05 (1H, s, 4-H)
4a	2.68	2.48	—	7.69-7.84 (5H, m, 2'-, 3'-, 4'-, 5'-, 6'-H); 7.97 (1H, dd, <i>J</i> ₇₈ = 8.8, <i>J</i> ₇₅ = 2.1, 7-H); 8.44 (1H, d, <i>J</i> ₈₇ = 8.8, 8-H); 8.90 (1H, d, <i>J</i> ₅₇ = 2.1, 5-H); 9.05 (1H, s, 4-H)
4b	2.70	2.49	—	7.77 (2H, d, <i>J</i> ₃₂ = 8.1, 3'-, 5'-H); 7.89 (2H, d, <i>J</i> ₂₃ = 8.1, 2'-, 6'-H); 7.97 (1H, d, <i>J</i> ₇₈ = 8.7, 7-H); 8.45 (1H, d, <i>J</i> ₈₇ = 8.7, 8-H); 8.91 (1H, s, 5-H); 9.09 (1H, s, 4-H)
5a	2.55	(2.35)	4.43 (CH ₂)	7.08 (12H, m)
5b	(1.00, 2.98, <i>J</i> = 7.1)	—	5.05 (CH ₂)	7.50-7.25 (13H)
6a*	—	2.31	4.51 (CH ₂); 6.56 (NH)	7.20-7.48 (9H, m); 7.51 (1H, s, 4-H); 8.00 (1H, d, <i>J</i> = 8.4, 8-H); 8.20 (1H, d, <i>J</i> = 8.4, 5-H)
6b	—	—	3.80 (d, <i>J</i> = 5.4, CH ₃); 5.65 (q, NH)	6.85 (1H, s, 2-H); 7.27-7.54 (5H, m); 7.80-7.82 (2H, m, 6-, 7-H); 7.85 (1H, s, 4-H); 8.10 (1H, d, <i>J</i> ₈₇ = 8.4, 8-H); 8.36 (1H, d, <i>J</i> ₅₆ = 8.4, 5-H)
6c	—	—	3.45 (2H, m, CH ₂ NH); 3.68 (2H, m, CH ₂ OH); 4.95 (t, <i>J</i> = 5.6, OH); 5.49 (t, <i>J</i> = 5.5, NH)	6.96 (1H, s, 2-H); 7.30-7.60 (5H, m); 7.75-7.86 (2H, m, 6-, 7-H); 7.90 (1H, s, 4-H); 8.10 (1H, d, <i>J</i> ₈₇ = 8.4, 8-H); 8.38 (1H, d, <i>J</i> ₅₆ = 8.4, 5-H)
6d	—	[2.15]	4.55 (2H, d, <i>J</i> = 6.3, CH ₂); 6.25 (t, NH)	6.15 (1H, s, 4-H _{furan}); 6.25 (1H, s, 3-H _{furan}); 7.30-7.59 (7H, m, H _{phenyl}); 7.59 (1H, s, 5-H _{furan}); 7.79 (1H, s, 4-H); 7.90 (1H, d, <i>J</i> ₈₇ = 8.4, 8-H); 8.22 (1H, d, <i>J</i> ₅₆ = 8.4, 5-H)
6e	—	(2.49)	4.57 (2H, d, <i>J</i> = 6.3, CH ₂); 6.25 (t, NH)	6.38 (2H, m, 3-, 4-H _{furan}); 7.03 (1H, s, 2-H); 7.31-7.49 (5H, m, H _{phenyl}); 7.59 (1H, s, 5-H _{furan}); 7.74 (1H, s, 4-H); 7.76 (1H, d, <i>J</i> ₇₈ = 7.4, 7-H); 7.88 (1H, d, <i>J</i> ₈₇ = 7.4, 8-H); 8.22 (1H, s, 5-H)

* Spectra of hydrochlorides.

product of addition of amine at the position 1 of N-(*p*-tolyl)amide of 2-[(1-benzylimino)ethyl]-5-methylbenzothiophene-3-acetic acid (**8**). Formation of such intermediate ketimines was postulated earlier by Dimroth and other researchers [1, 4-6].



EXPERIMENTAL

The ^1H NMR spectra were taken on a Gemini-200 (200 MHz) spectrometer in DMSO-d_6 , internal standard TMS. The characteristics of the synthesized compounds are given in Tables 1 and 2.

1,3-Disubstituted benzothieno[2,3-*c*]pyrylium perchlorates **1a,b**, **2a-c** were obtained by the procedures in [7, 8].

Synthesis of N-Aryl-1,3-dimethylbenzothieno[2,3-*c*]pyridinium Perchlorates (3) (General Procedure). The corresponding pyrylium perchlorates **1a,b** (5 mmol) were added to solution of arylamine (5 mmol) in acetic acid. The mixture was heated on a water bath for 1 h. The mixture was cooled down; the precipitate was filtered off and washed with alcohol and then ether, and then dried. Compounds **3b,c**, **4a,b** were obtained. They were crystallized from methanol.

Reaction of Benzothieno[2,3-*c*]pyrylium Perchlorates with Aliphatic Amines. The corresponding primary amine (5.1 mmol) was added to suspension of the substituted pyrylium perchlorates (5 mmol) in alcohol and then heated for 1 h and cooled down. The precipitate of pyridinium perchlorates **3a**, **5a,b** formed was filtered off, washed with acetone and then ether, and then dried. The obtained products were crystallized from methanol. The filtrate was evaporated down and then diluted with water; the precipitate of 1- R^4 -aminodibenzothiophenes **6a-e** was filtered off and then crystallized from isopropyl alcohol.

N-(*p*-Tolyl)amide of 2-[(1-Benzylimino)ethyl]-5-methylbenzothiophene-3-acetic Acid (8). Benzylamine (2.5 ml) was added to suspension of 1,6-dimethyl-3-(4'-tolylamino)benzothieno[2,3-*c*]pyrylium perchlorate **7** (5 mmol) in isopropyl alcohol (25 ml). The reaction mixture was cooled down and water (25 ml) was added. The precipitate was filtered off and washed with water, and then crystallized from a mixture of benzene-hexane, 1:3. ^1H NMR spectrum, δ , ppm (*J*, Hz): 2.15 (3H, s, 4'- CH_3); 2.42 (3H, s, 5- CH_3); 2.49 (3H, s, $\text{CH}_3\text{C}=\text{N}$); 4.13 (2H, s, CH_2CO); 4.79 (2H, s, CH_2N); 6.91 (2H, d, *J* = 8.0, 3'-, 5'-H); 7.03 (2H, d, *J* = 8.0, 2'-, 6'-H); 7.24-7.31 (5H, m, H); 7.46 (1H, d, *J* = 8.2, 6-H); 7.80 (1H, d, *J* = 8.2, 7-H); 7.86 (1H, s, 4-H); 10.48 (1H, s, NH).

REFERENCES

1. K. Dimroth, *Angew. Chem.*, **72**, 331 (1960).
2. G. N. Dorofeenko, E. I. Sadekova, and V. V. Goncharova, *Khim. Geterotsikl. Soedin.*, 1308 (1970).
3. E. V. Kuznetsov and G. N. Dorofeenko, *Khim. Geterotsikl. Soedin.*, 1437 (1971).
4. G. P. Safaryan, I. V. Shcherbakova, G. N. Dorofeenko, and E. V. Kuznetsov, *Khim. Geterotsikl. Soedin.*, 1608 (1981).
5. S. V. Verin, D. E. Tosunyan, and E. V. Kuznetsov, *Khim. Geterotsikl. Soedin.*, 1468 (1991).
6. V. I. Terenin, L. G. Yudin, R. S. Sagitullin, V. N. Torocheshnikov, V. I. Dulenko, Yu. A. Nikolyukin, and A. N. Kost, *Khim. Geterotsikl. Soedin.*, 73 (1983).
7. V. I. Dulenko, S. V. Tolkunov, and N. N. Alekseev, *Khim. Geterotsikl. Soedin.*, 1351 (1981).
8. S. V. Tolkunov, *Khim. Geterotsikl. Soedin.*, 1335 (1998).